

HiAl Biofuels for CHP Plants

**Reduced Emissions and Cost reduction in the
Combustion of High Alkali Biofuels**

Klaus Hilpert

**Research Center Jülich,
52425 Jülich, Germany**



**BIO-ENERGY
ENLARGED PERSPECTIVES**

Budapest ,16-17 October 2003

Technical aspects of HIAL biofuel combustion - problems to be solved -

Grate, Fluidized bed, Suspension

High temperature corrosion of heat exchangers

High slagging and fouling propensity

Relatively high emissions of NO_x , SO_2 and HCl

**current emissions for grate: $300 \text{ mg/Nm}^3 \text{ SO}_2$
 $200 \text{ mg/Nm}^3 \text{ HCl}$
 $400 \text{ mg/Nm}^3 \text{ NO}_x$**

Aims of the Project

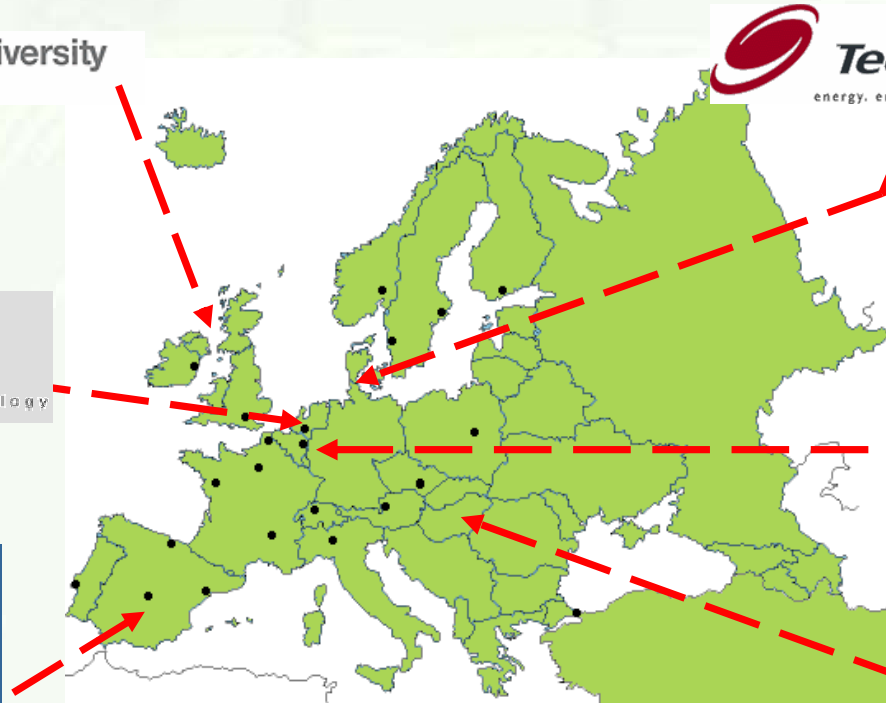
- **Fundamental understanding of combustion and release**
 - **Grate firing**
 - **Suspension firing**
 - **Fluidised-bed combustion**
 - **Mass spectrometry**
 - **Modelling**
- **Reduced SO₂ emissions in grate firing by >33% (<200 mg/Nm³)**
- **New designs for grate-fired CHP plants with lower installation costs and running costs**



Consortium of the Project

CITY City University
London
Suspension firing

TU Delft
Delft University of Technology
CFBC



Tech-wise
energy. environment. knowledge.

DTU
Grate firing



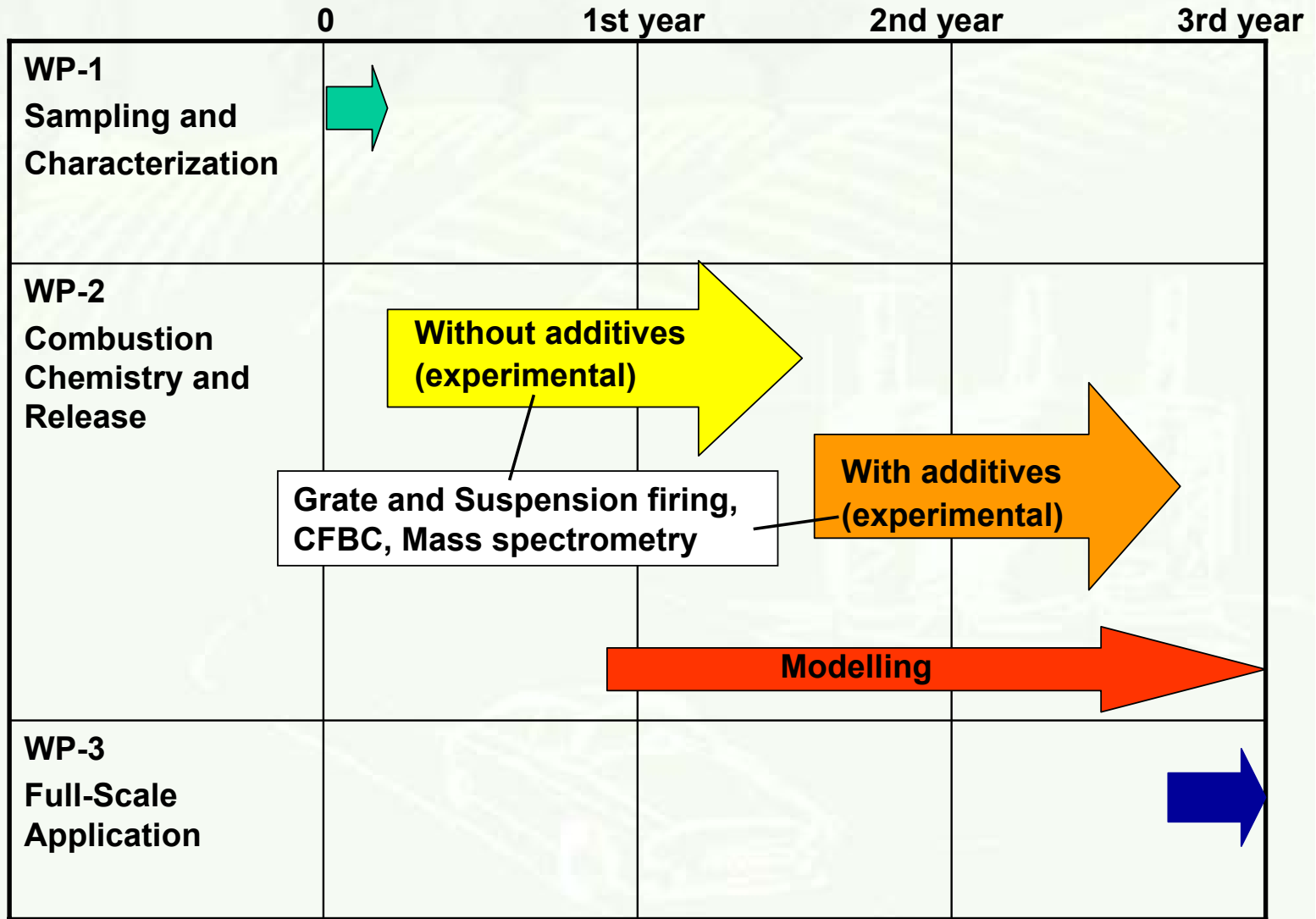
Mass spectrom.



Modelling



Timetable



Samples

Spain:

HIAL 1 - Rice
 HIAL 2 - Oats
 HIAL 3 - Wheat Marius
 HIAL 4 - Rape
 HIAL 5 - Winter barley
 HIAL 6 - Wheat Soisson
 HIAL 7 - Carinata
 HIAL 8 - Spring barley
 HIAL 9 - Maize

Denmark:

HIAL 10 - Wheat 2000
 HIAL 11 - Wheat 2001
 HIAL 12 - Barley 2001

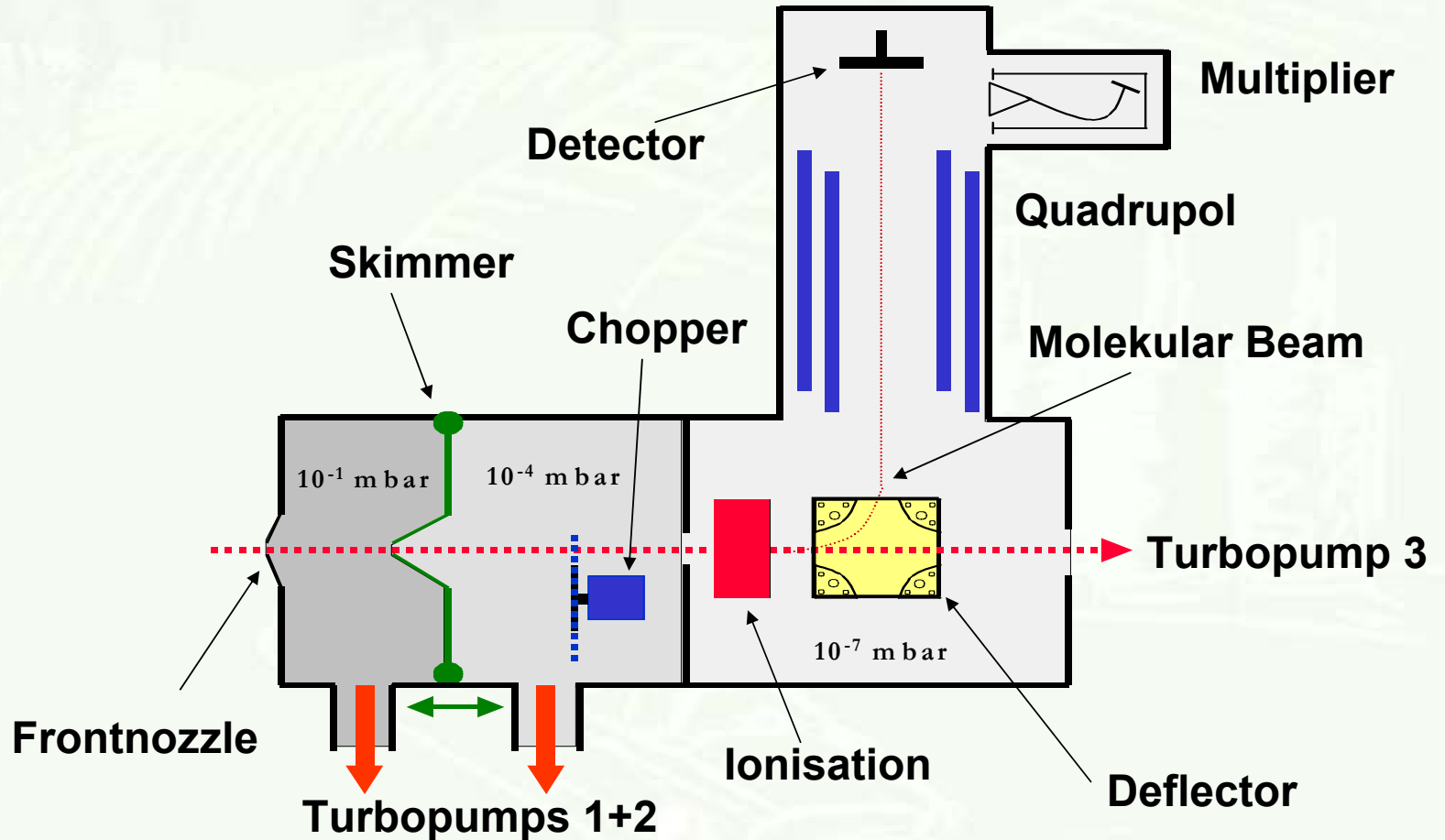
← **12 samples** →

↓ **4 key samples** ↓

<u>Sample</u>	<u>Straw type</u>	<u>Characteristics</u>	<u>K/Si-ratio</u>	<u>S/Cl-ratio</u>	<u>K/Cl-ratio</u>
HIAL 10	Danish wheat (2000)	Typical composition	0.9	0.5	4.3
HIAL 5	Spanish winter barley	High in K and Cl	3.0	0.2	2.2
HIAL 2	Spanish oats	Low in K, Cl and Si	2.1	2.7	10.2
HIAL 7	Spanish Carinata	High in K and S Low in Si and Cl	28	4.7	28

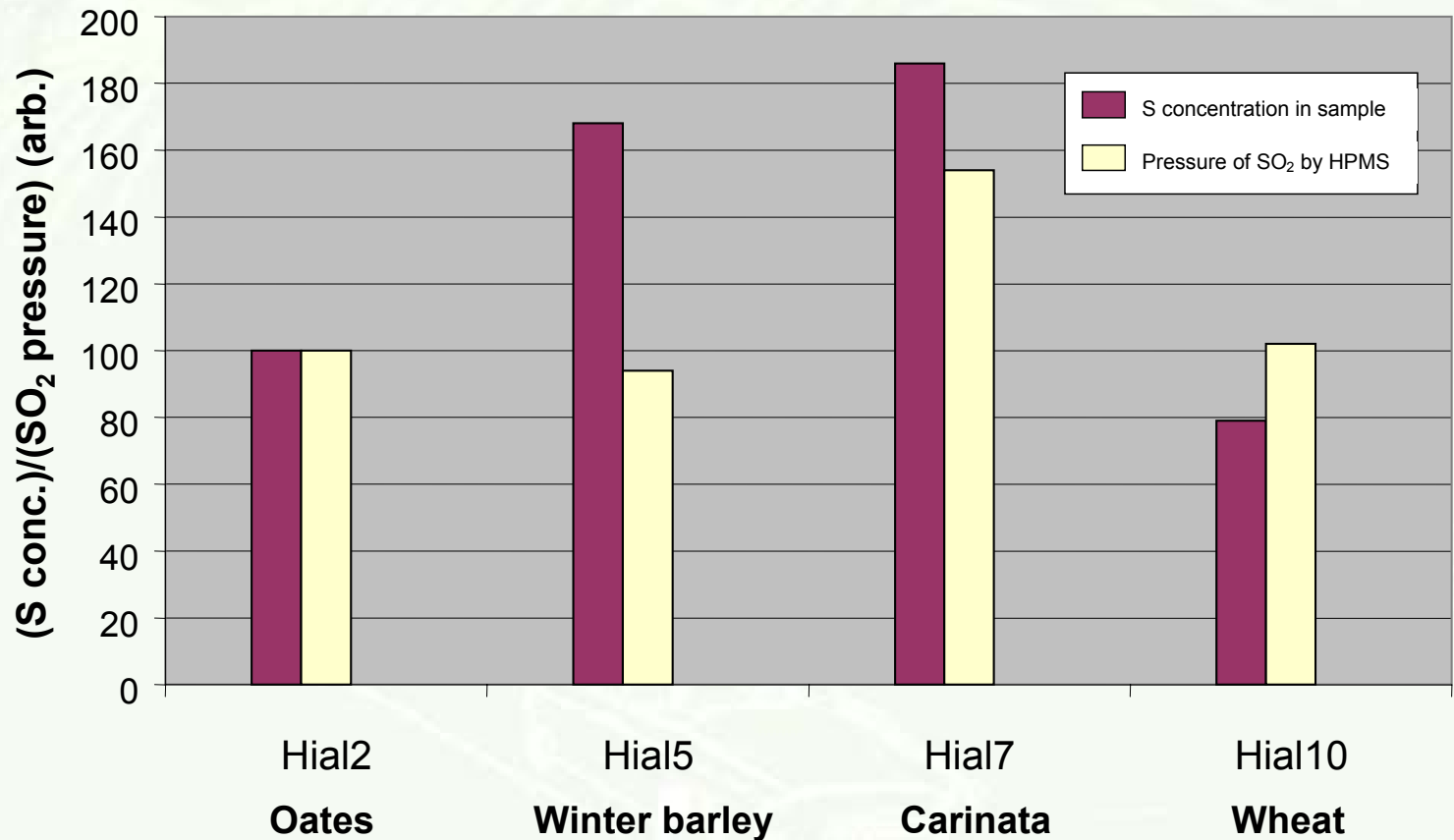
High Pressure Mass Spectrometer

(Research Centre Jülich)



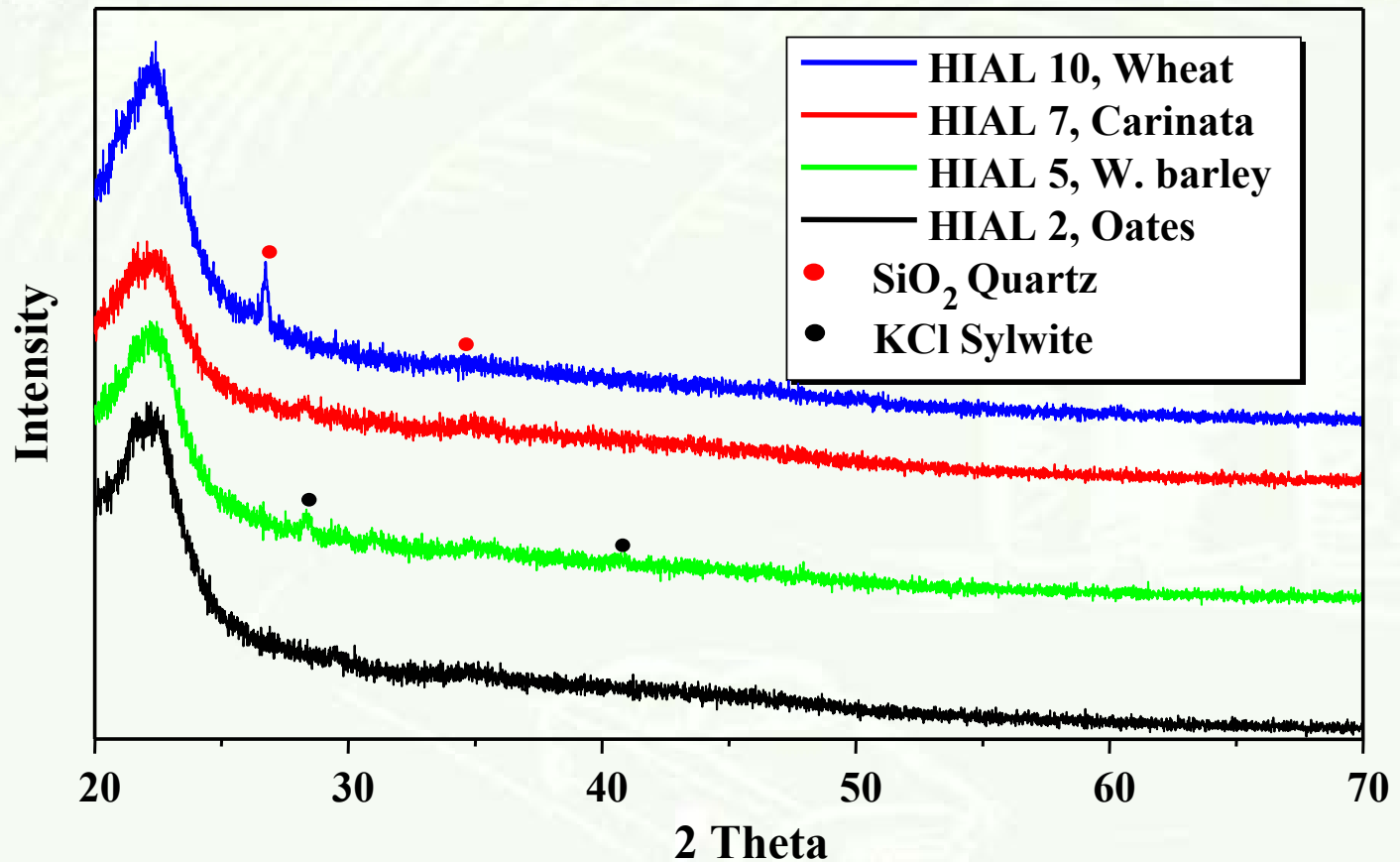
SO₂ Release at 800 °C and S Concentration in different Samples

(Res. Centre Jülich)



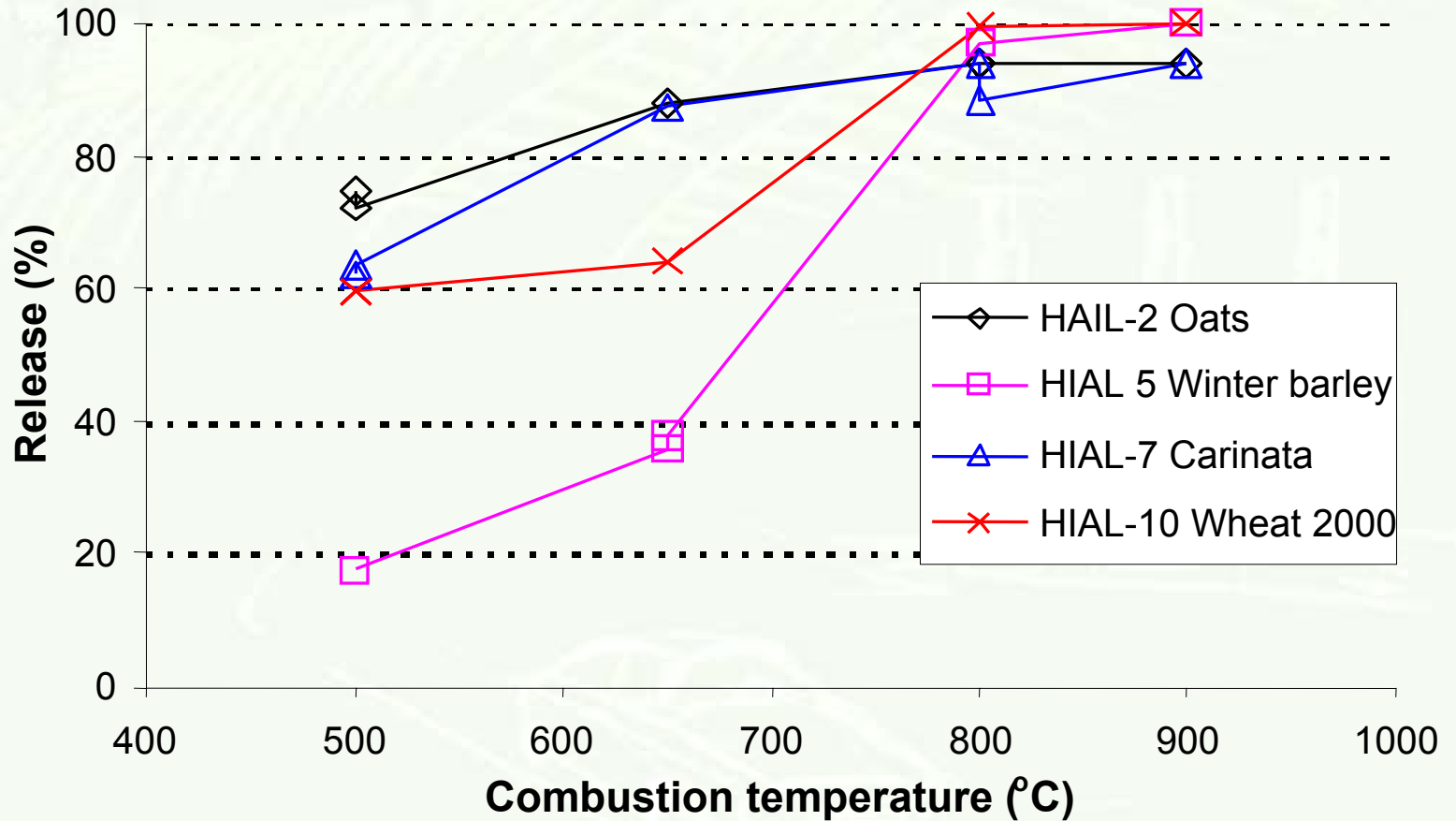
XRD of Biomass Samples

(Res. Center Jülich)



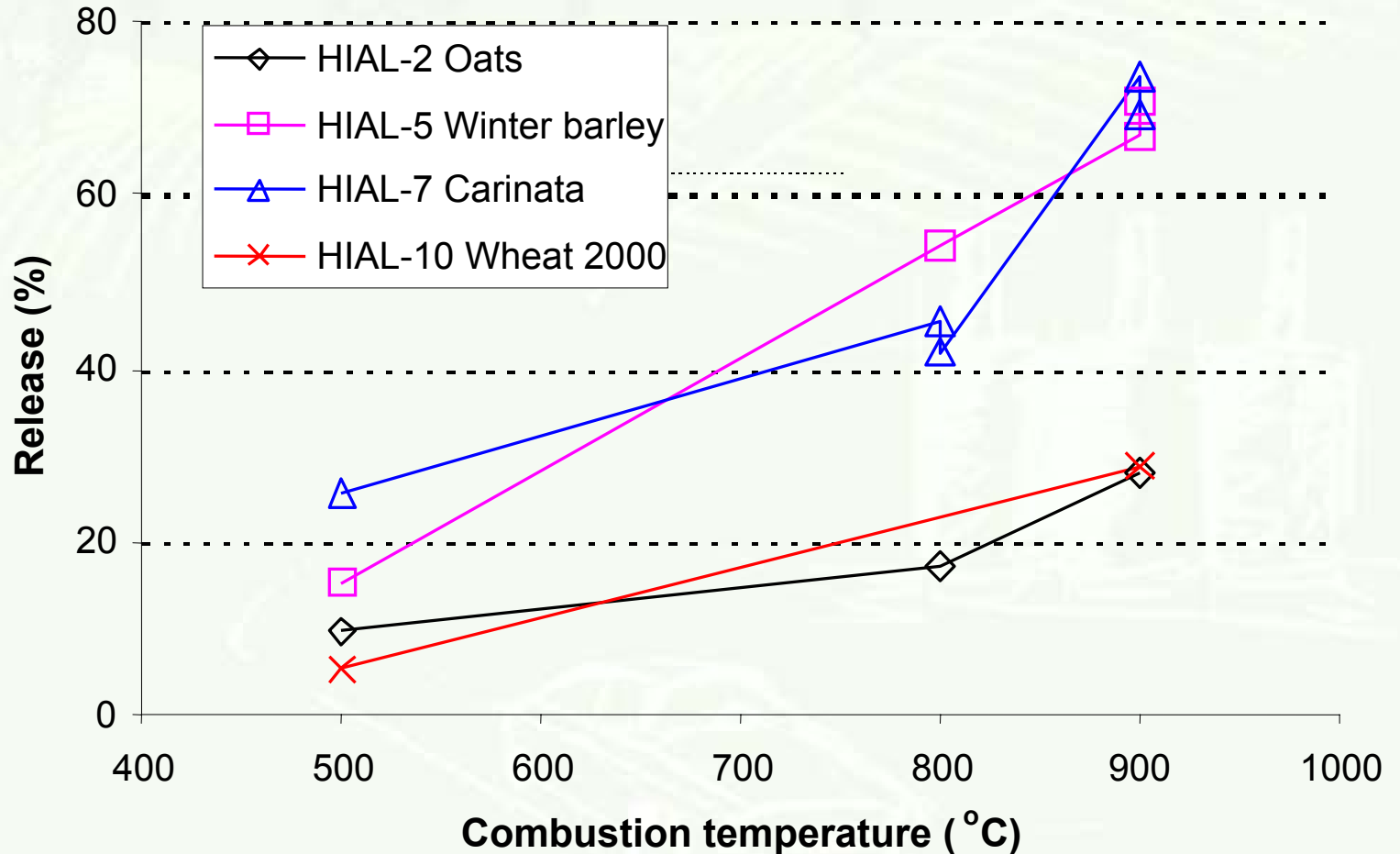
Chlorine Release During Combustion

(Danish Technical University)

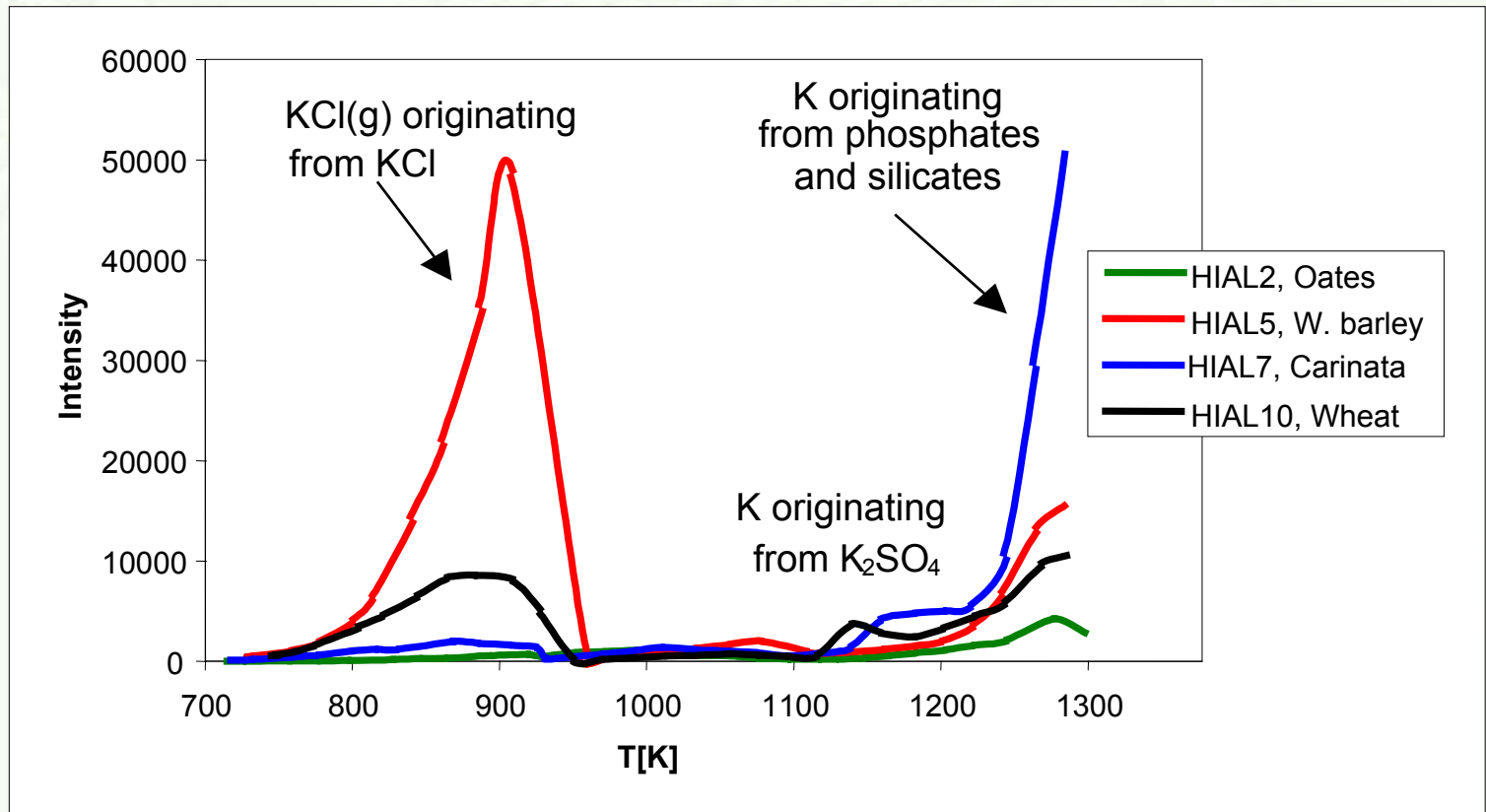


Potassium Release During Combustion

(Danish Technical University)



MS Vaporisation Measurements of Biomass/Coal (1:1) Ashes (550 °C) (Res. Center Jülich)



Results on the Release of Sulphur, Potassium and Chlorine

Sulphur

- High amounts of Potassium lower the SO₂ release by formation of Potassium sulphates
- Silica promotes the release of SO₂ by potassium capture inhibiting the K₂SO₄ formation

Potassium

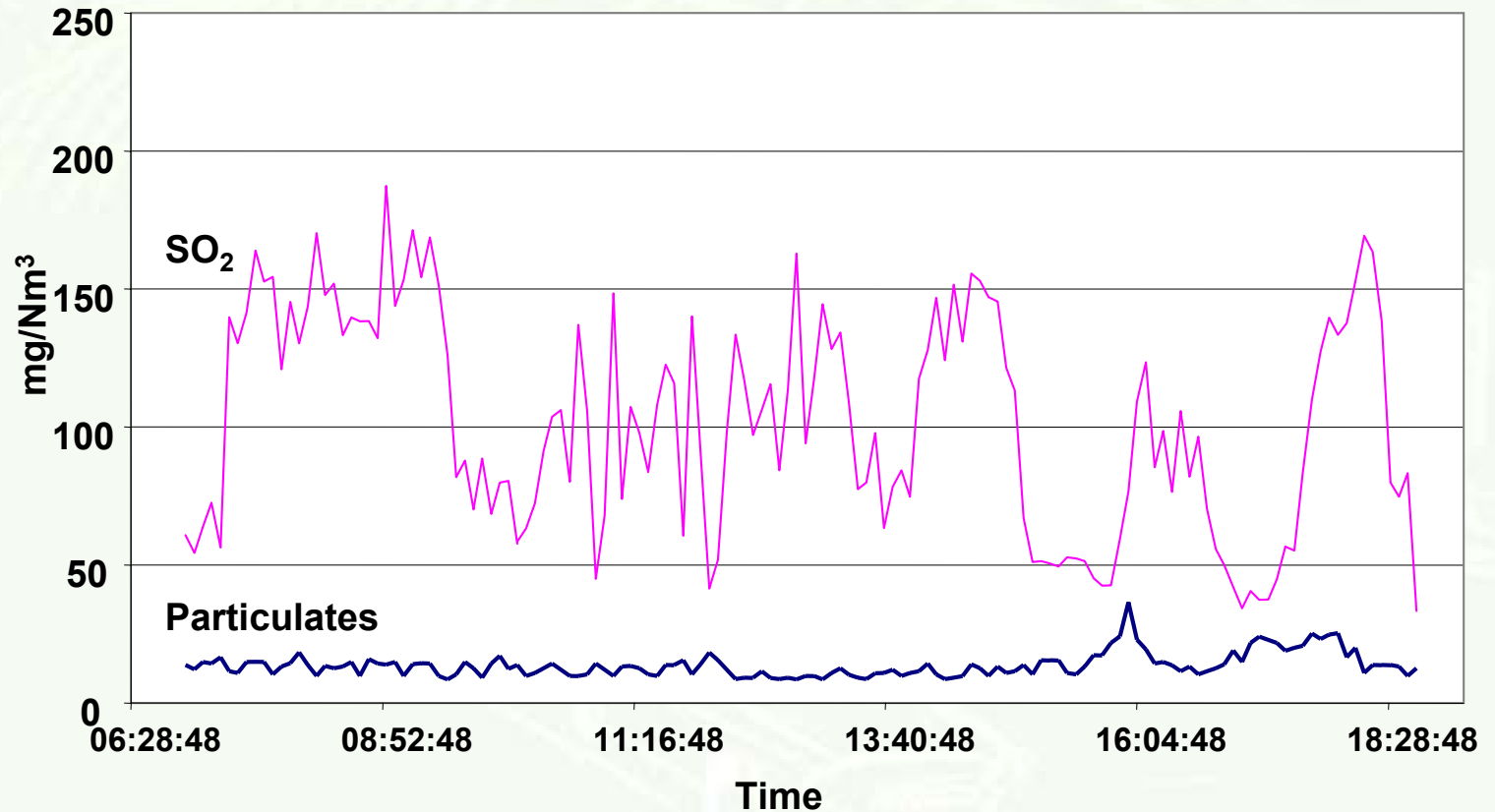
- Release of Potassium rather depends on Chlorine content than Potassium content of the samples
- High silica content reduces Potassium release

Chlorine

- Chlorine is completely released at $T \geq 800 \text{ }^{\circ}\text{C}$

Sulphur Release at Ensted Power Plant (TechWise)

Barley straw, 08.10.2002, 7:00-18:45



Timetable

