

Biomass Conversion in Supercritical Water

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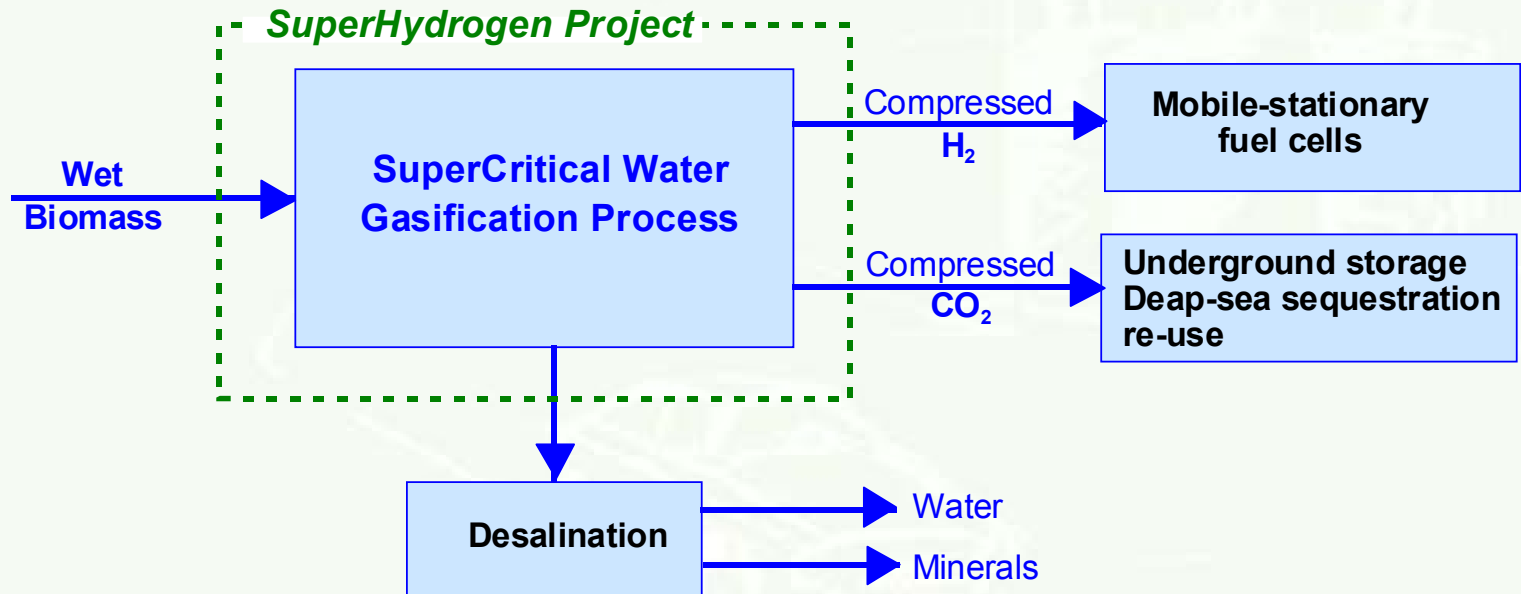


**BIO-ENERGY
ENLARGED PERSPECTIVES**

Budapest, 16-17 October 2003

Biomass and waste conversion in supercritical water for the production of renewable hydrogen

SuperHydrogen



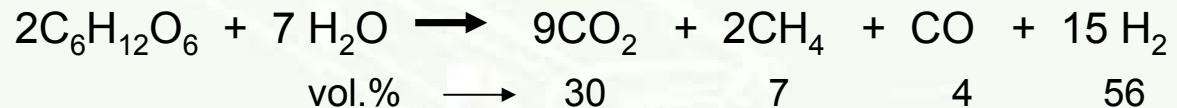
Gasification in Supercritical water

Supercritical water: $T > 374 \text{ }^\circ\text{C}$
 $P > 22 \text{ MPa}$
 one fluidium (no gas-liquid interface)

Why gasification in supercritical water ?

- Suitable for very wet biomass (moisture content $> 70 \text{ wt.}\%$)
- Produced gas rich in hydrogen ($> 50 \text{ vol.}\%$)
- Gas available at high pressure ($\sim 300 \text{ bar}$)
- Gas is rather clean (no minerals, tars etc)
- Enables counter-current heat exchange between feedstock and product

Example: reaction of glucose



Overall Objective of SuperHydrogen:

Development supercritical water gasification process for cost-effective (< 12 €/GJ) conversion of wet biomass/waste into hydrogen with energy efficiency exceeding 60%.

1. Feedstock selection, preparation & pressurizing

- Selection wet biomass/waste streams in Europe
- Development of process to produce pumpable slurries (up to 30 wt.%)
- Pump selection and testing



2. Development supercritical water gasification process

- Process fundamentals using micro-set-ups
- Pilot-plant tests with model components
- Pilot-plant tests with real biomass



3. *Product upgrading: development catalytic membrane reactor*

- Shift CO for at least 70%
- Reform methane for at least 70%
- Separate hydrogen (purity > 98 vol.%)



Dytech Ltd.

4. *Process Modelling*

- Kinetics and phase equilibria models
- Modules for unit operations
- Overall process model



5. *Basic engineering & cost estimate*

- Basic engineering of complete process from biomass to pure hydrogen
- Cost estimates (investment & operational)

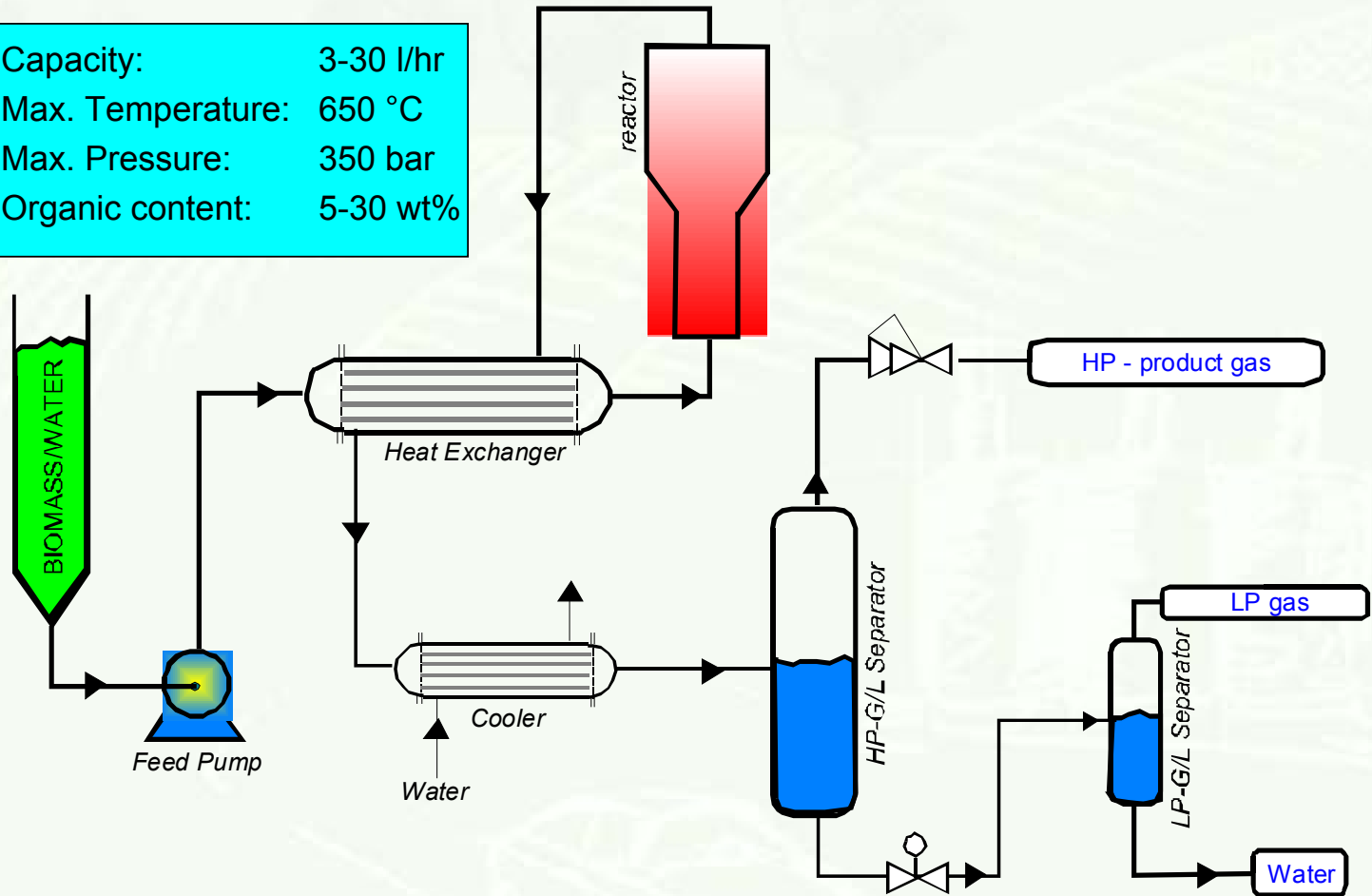


6. *Specification of H₂, safety aspects and alternatives*

- Specifications of H₂ for different end-uses
- Alternative processes for H₂ production
- Safety aspects of H₂ production



Capacity: 3-30 l/hr
Max. Temperature: 650 °C
Max. Pressure: 350 bar
Organic content: 5-30 wt%



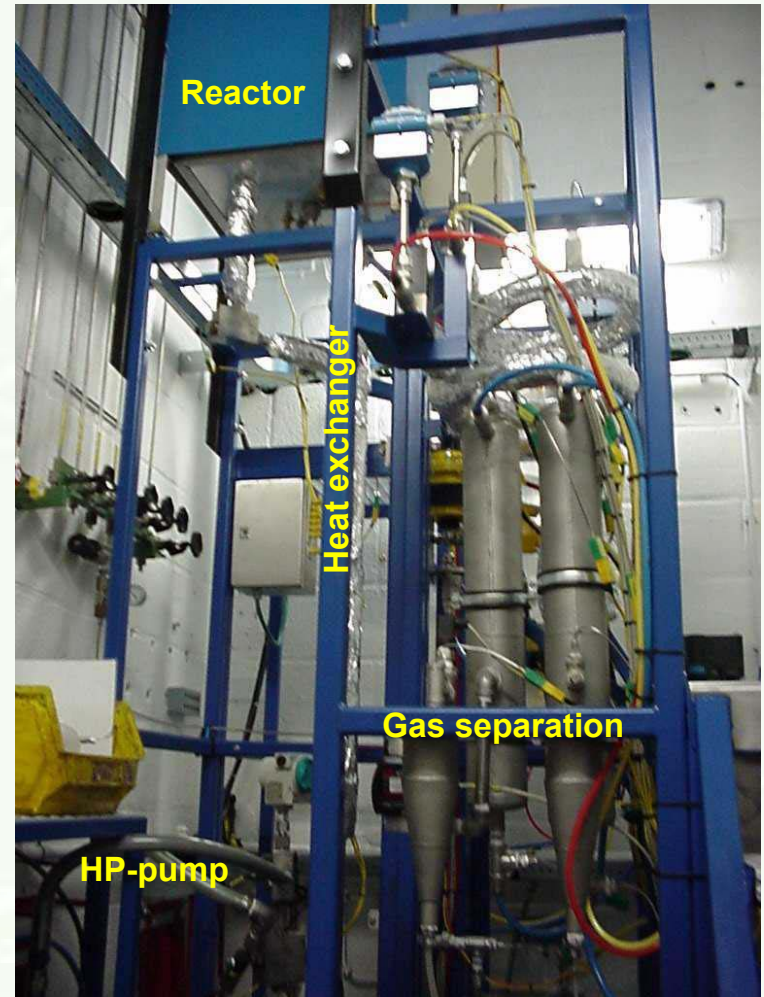
Basic flow diagram supercritical gasification process (pilot-plant)



Process control

Feeding section

Feeding section & process control



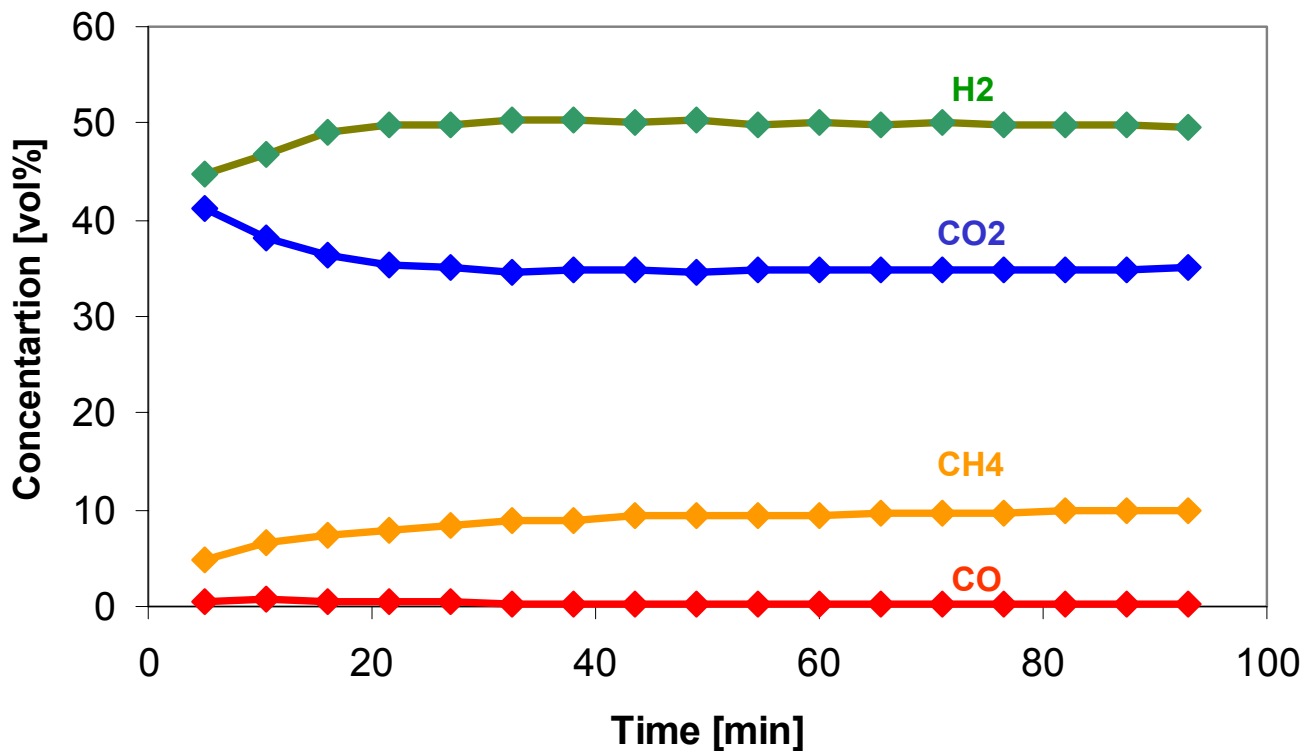
Reactor

Heat exchanger

Gas separation

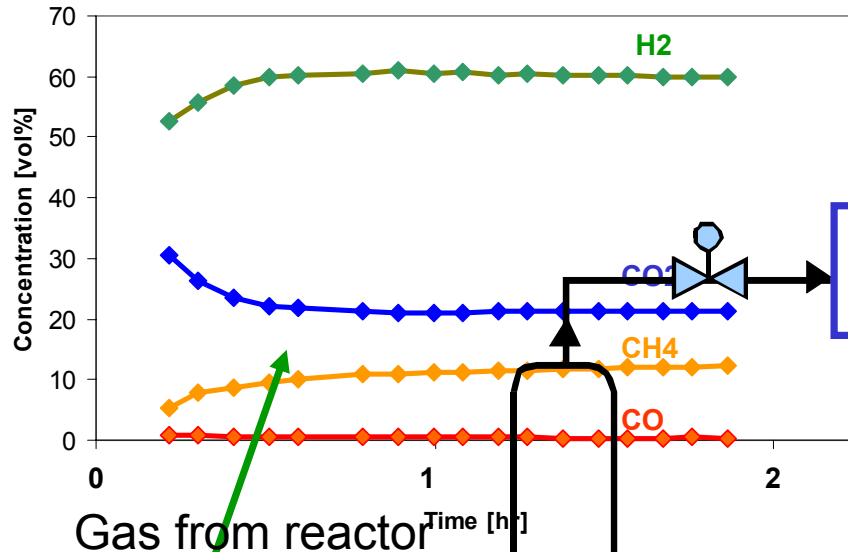
HP-pump

Reactor & gas separation



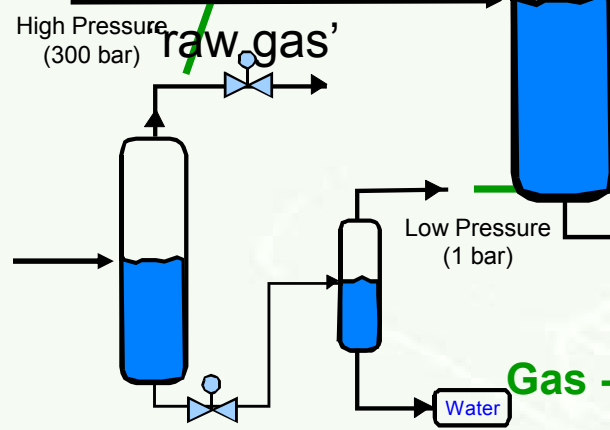
Concentration of main gas components after the reactor as a function of time on stream;

Feedstock: 5 wt.% glycerine, 0.0075 wt.% NaOH; flow = 7 kg/hr; T = 580 °C; P = 270 bar.

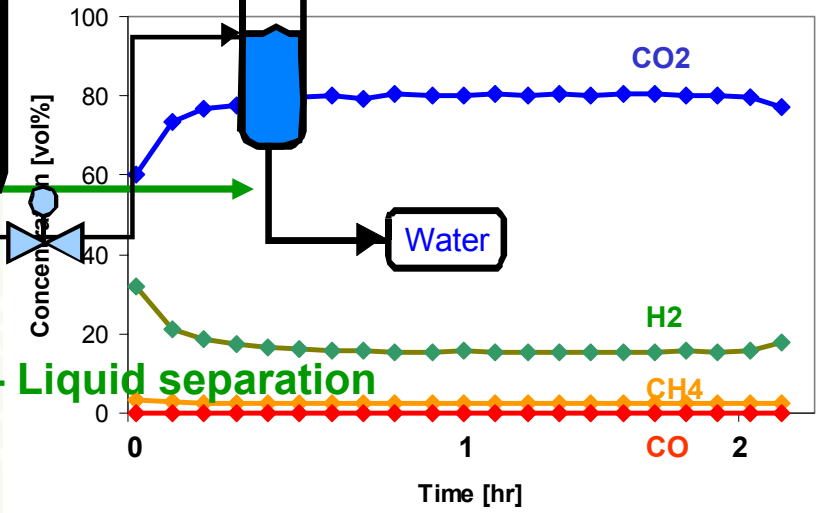


High Pressure Gas
(300 bar)

Low Pressure Gas
(1 bar)



Gas - Liquid separation



Gas - Liquid separation

Time [hr]



Typical gas concentrations (high pressure product gas)

- Gasification of 5 wt.% glycerine with and without additives
- Capacity: 7 kg/hr
- $T = 580\text{ }^{\circ}\text{C}$; $P = 270\text{ bar}$

Additives Compound (vol.%)	-	Na₂CO₃ 0.01 wt.%	NaOH 0.0075 wt.%
Hydrogen (H ₂)	29	59	60
Carbon Monoxide (CO)	30	1.5	0.4
Carbon Dioxide (CO ₂)	13	21	21
Methane (CH ₄)	15	13	12
C2 -	11	4.7	4.7
C3 -	2	1.0	1.5
LHV (MJ/Nm ³)	21.1	15.1	15.2

Future work

- 2004 Slurry preparation demonstrated and tested with pump
- 2003 - 2005 Fundamental research in capillaries / continuous micro set-up
- 2003 Pilot-plant tests with model components (glycerol, MeOH, starch)
- 2004 - 2005 Pilot-plant tests with real biomass
- 2004 Prototype gas upgrading reactor to be tested with artificial gas
- 2005 Integration upgrading reactor in pilot plant and integrated testing
- 2003 Preliminary techno-economic evaluation overall process
- 2005 Final techno-economic evaluation

More info: <http://superh2.grensy.info>